

Biomimetic Asymmetric Synthesis of (*R*)-GTRI-02 and (3*S*,4*R*)-3,4-Dihydroxy-3,4-dihydronaphthalen-1(2*H*)-ones

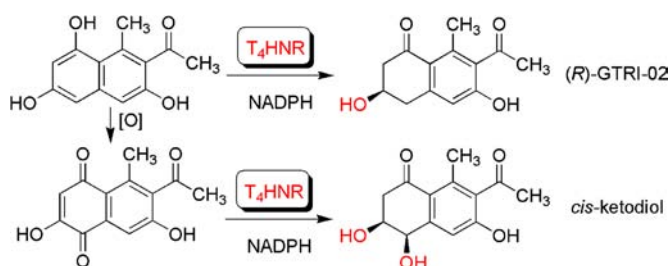
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ABSTRACT



The NADPH-dependent tetrahydroxynaphthalene reductase (T₄HNR) from *Magnaporthe grisea* was used for the biomimetic synthesis of (*R*)-GTRI-02 by stereoselective reduction of 1-(3,6,8-trihydroxy-1-methylnaphthalen-2-yl)ethanone. This also led to the isolation of a (3*S*,4*R*)-*cis*-ketodiol formed by T₄HNR-catalyzed reduction of the corresponding hydroxynaphthoquinone. Flaviolin and lawsone also reduced to corresponding *cis*-ketodiol in good yields.

Naphthol reductases belong to the large family of short-chain dehydrogenases/reductases (SDR) and show a unique ability to catalyze asymmetric NADPH-dependent reduction of polyhydroxynaphthalenes.¹ Although the use of naphthol reductases as biocatalysts in synthesis looks promising, so far their application has remained limited to the reduction of only a few physiological substrates.^{1–4} Tetrahydroxynaphthalene reductase (T₄HNR) from *Magnaporthe grisea* has been used by us and others to catalyze the reduction of 1,3,6,8-tetrahydroxynaphthalene (T₄HN, **1**) to (*R*)-scytalone (**2**) in 33% yield (Scheme 1, A).^{1,4,5} T₄HNR is

one of the two naphthol reductases involved in the biosynthesis of dihydroxynaphthalene melanin.^{2–4} For naphtholic substrates to be reduced by T₄HNR, the 1,3-dihydroxy substitution pattern represents the essential structural motif.¹ Herein, we report the chemoenzymatic synthesis of the natural product GTRI-02 (**3**) using T₄HNR as well as the unexpected recognition of hydroxynaphthoquinones as substrates which are reduced by T₄HNR to *cis*-ketodiol.

The putatively polyketidic GTRI-02 (**3**), isolated from soil actinomycetes *Micromonospora* sp.,⁶ has been shown to possess antioxidant properties. More recently, it has also been extracted from *Streptomyces* strain GW4184 and *Streptomyces* sp. ANK313.^{7,8} We proposed that **3** is biosynthesized via an enzymatic reduction.

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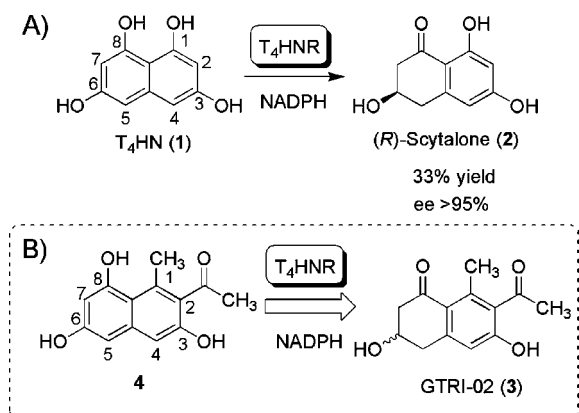
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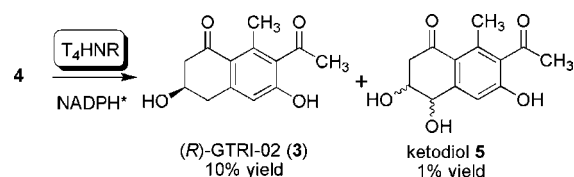
Scheme 1. (A) T₄HNR-Catalyzed Reduction of its Physiological Substrate T₄HN (1). (B) Retrosynthetic Proposal for the Biomimetic Enzymatic Synthesis of GTRI-02 (3)



The biomimetic retrosynthetic analysis of 3 guided us to the acetylated trihydroxynaphthalene 4 as the required substrate (Scheme 1, B). Compared to T₄HN (1), naphthol 4 contains an additional acetyl group, in principle a potentially better substrate unit to be reduced by oxidoreductases.

Compound 4 was synthesized in four straightforward steps starting from 3,5-dimethoxyphenylacetic acid in an overall yield of 25% (see the Supporting Information). Naphthol 4 was then reduced with T₄HNR. NADPH was used as a cofactor and regenerated using glucose and glucose dehydrogenase (GDH) in all enzyme-catalyzed reactions. The transformation proceeded as proposed and resulted in the formation of GTRI-02 (3) as the major product after 24 h (17% conversion), supporting the argument for the putative involvement of a similar naphthol reductase in the biosynthesis of 3.⁹ Unexpectedly, ketodiol 5 was obtained as a side product in the same transformation (Scheme 2).

Scheme 2. T₄HNR-Catalyzed Reduction of 4



* NADPH was regenerated using glucose/glucose dehydrogenase.

GTRI-02 (3) was isolated and purified by preparative HPLC (RP-18) to provide pure material in 10% isolated yield. The absolute configuration was determined as (R) by using circular dichroism (CD). Additionally, the

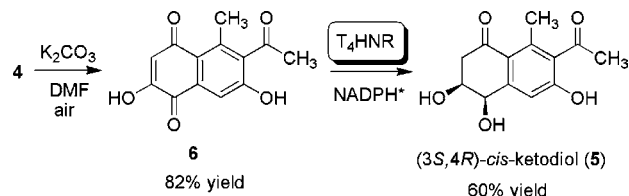
(9) 2-Acetyl-1,3,6,8-tetrahydroxynaphthalene (a) and 6-hydroxymyriszsin (b) were synthesized and tested with T₄HNR; however, the transformations did not result in any reduction. (a) Wheeler, M. H.; Abramczyk, D.; Puchhaber, L. S.; Naruse, M.; Ebizuka, Y.; Fujii, I.; Szaniszlo, P. J. *Eukaryotic Cell* **2008**, *7*, 1699–1711. (b) Harris, T. H.; Wittek, P. J. *J. Am. Chem. Soc.* **1975**, *97*, 3270–3271.

bis-*p*-bromobenzoate derivative of 3 was prepared, and CD spectroscopy, according to the method of Harada and Nakanishi (exciton coupling), was used to unambiguously verify the assignment.¹⁰ Since both the natural substance⁶ and the enzymatically synthesized product show negative specific rotation, the absolute configuration of the former must also be (R).

The unexpected side product of the enzymatic reduction, ketodiol 5, was isolated in 1% yield by column chromatography. Further analysis of the T₄HNR-catalyzed reduction reaction of 4 led to the detection of the presence of 2-hydroxy-1,4-naphthoquinone 6, possibly formed by nonenzymatic aerobic oxidation. Although the transformation was performed under N₂, incomplete removal of oxygen from the buffer solution might account for the oxidation of compound 4. We assumed that 2-hydroxy-1,4-naphthoquinone 6 could have been reduced to diol 5 by T₄HNR, using 2 equiv of NADPH.

In order to prove this assumption, 6 was prepared in 82% isolated yield by the oxidation of compound 4 using K₂CO₃ in DMF.¹¹ Hydroxynaphthoquinone 6 was then employed as a substrate and was reduced by T₄HNR to exclusively give the above-mentioned vicinal diol 5 with 80% conversion (60% yield, dr *cis/trans* ≥ 99:1) (Scheme 3). The relative configuration of 5 was elucidated by single-crystal X-ray analysis of the 4-biphenylboronic ester derivative. The absolute (3*S*,4*R*)-configuration was determined by vibrational circular dichroism (VCD) and quantum chemical calculations (Gaussian 09¹²) (see the Supporting Information).

Scheme 3. T₄HNR-Catalyzed Reduction of Hydroxynaphthoquinone 6



T₄HNR-catalyzed reduction of 2-hydroxy-1,4-naphthoquinone 6 to *cis*-ketodiol 5 also hints at the possible

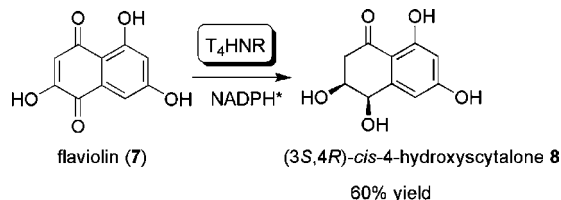
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involvement of T₄HNR or related enzymes in the putative biosynthetic reduction of flaviolin (**7**) to 4-hydroxyscytalone (**8**).¹³ To support our argument, **7** was synthesized in 80% yield by the oxidation of T₄HN (**1**)¹¹ and was used as a substrate for T₄HNR (Scheme 4).

Scheme 4. T₄HNR-Catalyzed Reduction of Flaviolin (**7**)

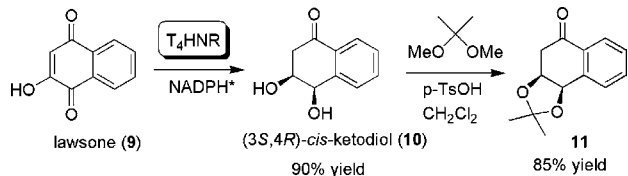


The transformation was performed using glucose/GDH for cofactor regeneration and resulted in the formation of *cis*-4-hydroxyscytalone (**8**) with quantitative conversion and high diastereoselectivity ($dr_{cis/trans} = 99:1$) (determined by ¹H NMR spectroscopy). A lower isolated yield of 60% was obtained due to decomposition of **8**.

Relative to transformations performed in the presence of air, much better yields were obtained when the transformation was performed under N₂ and after degassing. *cis*-4-Hydroxyscytalone (**8**) has been isolated previously from various sources and has also been proposed to be formed during melanin biosynthesis of *Wangiella dermatitidis* along with other naphthalene-based metabolites.^{9a,13–15}

To stress hydroxynaphthoquinones as general substrates for T₄HNR, another natural product, lawsone (**9**), was used as a substrate (Scheme 5). This transformation resulted in quantitative reduction, and the *cis*-ketodiol **10** was isolated in 90% yield.

Scheme 5. Reduction of Lawsone (**9**) Catalyzed by T₄HNR



The same transformation was performed in up to a 100-mg scale with no significant change in the yield of ketodiol **10**. The configuration of **10** was determined by VCD of its acetonide **11** and assigned as (3*S*,4*R*) (Figure 1).

For *cis*-4-hydroxyscytalone (**8**), the absolute configuration (3*S*,4*R*) was assigned by comparison of the CD spectra of **8** and **10** (see the Supporting Information).

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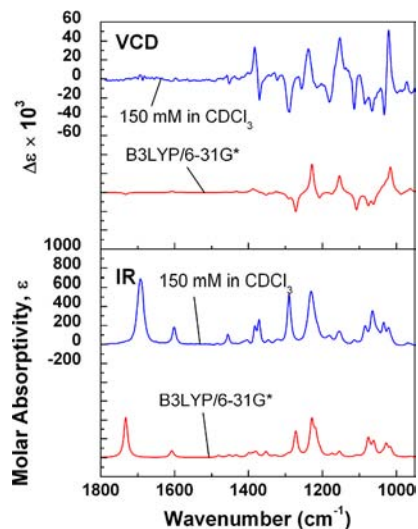


Figure 1. Experimental IR and VCD spectra of the acetonide **11** (150 mM in CDCl₃) and the theoretical IR and VCD spectra calculated¹² (B3LYP/6-31G* level) for the (3*S*,4*R*)-configured acetonide **11**. Good agreement in frequencies and sign between the calculated and observed spectra allows for assignment of the absolute configuration.

Although several studies^{13b} have proposed that *cis*-4-hydroxyscytalone (**8**) is formed by the reduction of flaviolin (**7**), the absolute configuration of **8** and the enzymatic reduction of flaviolin by T₄HNR have now been established for the first time.

Acceptance by T₄HNR of different hydroxynaphthoquinones as substrates and their reduction to homochiral *cis*-ketodiols hint at further biosynthetic considerations: we propose that in the biosynthesis of 3-hydroxy-3,4-dihronaphthalen-1(2*H*)-ones (and the corresponding anthracenones), e.g., GTRI-02 (**3**), T₄HNR-like enzymes might be involved. Characterization of these enzymes with a probably expanded substrate range will broaden the

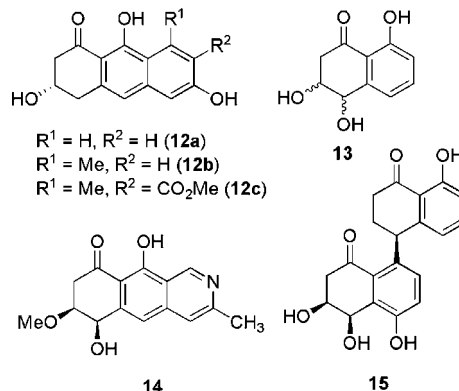


Figure 2. Natural products containing 3-hydroxy-3,4-dihydroanthracen-1(2*H*)-one (**12a–c**) and 3,4-dihydroxy-3,4-dihydroanthracen-1(2*H*)-one substructures (**13–15**).

scope of asymmetric “phenol”-reducing methods. Moreover, the biosynthesis of 3,4-dihydroxy-3,4-dihydronaphthalen-1(2*H*)-ones (and the corresponding anthracenones) might proceed through T₄HNR-like transformations, as shown above for the reduction of hydroxy-1,4-naphthoquinones **6**, **7**, and **9**.

Several natural products with similar substructures show interesting biological activities. For example, **12a** isolated from *Dendrobium* sp.¹⁶ and **12b** and **12c** isolated from the roots of *Lomatophyllum*¹⁷ possess the 3-hydroxy-3,4-dihydroanthracen-1(2*H*)-one substructure (Figure 2).

cis-Ketodiols contain a 3,4-dihydroxy-1-tetralone substructure which is also a part of several natural products. Such an example is 3,4,8-trihydroxy-1-tetralone (**13**) which has been isolated from several fungal species and shows antifungal activity.^{18–20} Chrysanthone A (**14**)²¹ and

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cladosporol D (**15**)²² are further examples of natural products with a vicinal *cis*-diol containing a similar substructure.²³

In summary, T₄HNR was successfully applied for the chemoselective reduction of naphtholic substrates; no reduction of the acetyl side chain present in compounds **4** and **6** was observed. We have shown that the enzymatic reduction of aromatic hydroxy groups and, especially, 2-hydroxy-1,4-naphthoquinones can be used for the asymmetric synthesis of related compounds.²³ Moreover, the discussed biosynthetic considerations will help to identify related enzymes.

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Supporting Information Available. Experimental procedures and full spectroscopic data for all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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The authors declare no competing financial interest.